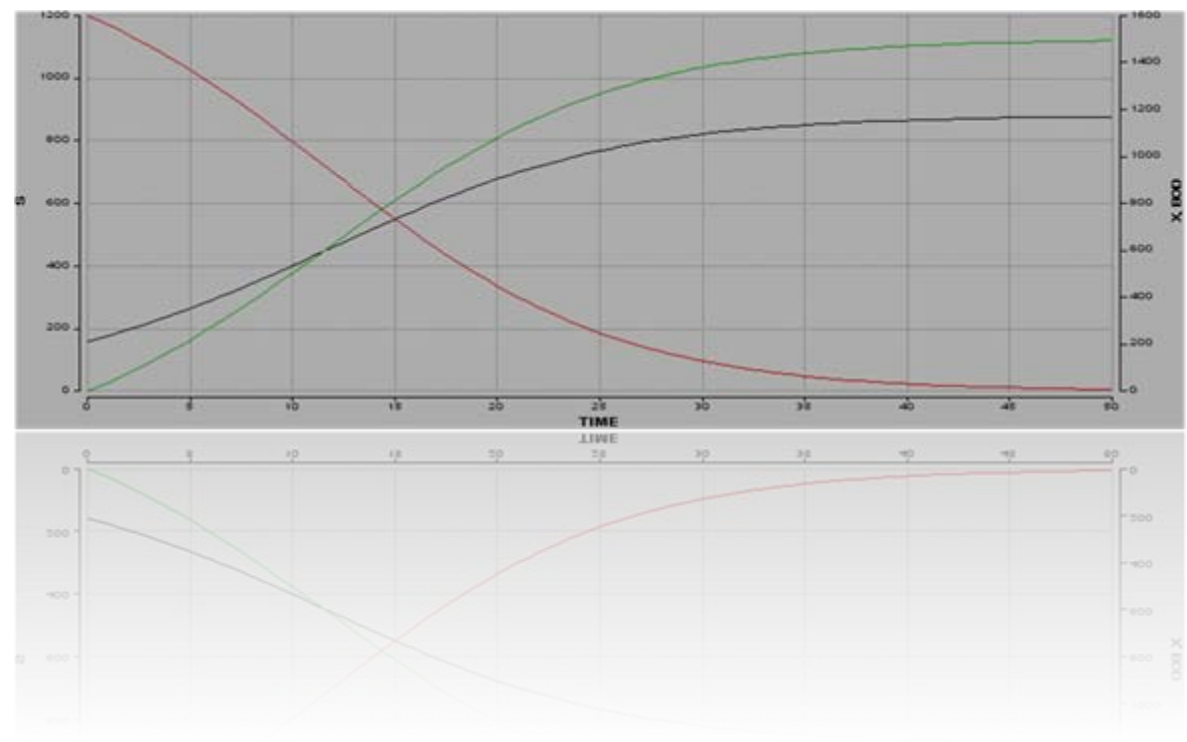


# Modelling Aliphatic Polyesters Biodegradation

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# Biodegradable Plastics

## **NATURAL BIODEGRADABLE PLASTICS**

- Based on renewable resources
- Can be either naturally produced or synthesised from renewable resources

## **NON-RENEWABLE SYNTHETIC BIODEGRADABLE PLASTICS**

- Petroleum-based

**BIODEGRADABLE PLASTICS ARE BLENDED WITH SYNTHETIC POLYMERS TO PRODUCE PLASTICS WHICH MEET FUNCTIONAL REQUIREMENTS**

**RANGE OF  
POTENTIAL  
APPLICATIONS**

**MAJOR POTENTIAL  
DISPOSAL  
ENVIRONMENTS**

**SOME ADVERSE  
ENVIRONMENTAL  
RISKS**

## RANGE OF POTENTIAL APPLICATIONS

- Film including overwrap, shopping bags, waste and bin liner bags, composting bags, mulch film, silage wrap, landfill covers, packaging - including O<sub>2</sub> and H<sub>2</sub>O barriers, bait bags, nappy backing sheet, and cling wrap
- Flushable sanitary products
- Sheet and non woven packaging
- Bottles
- Liquid paper board
- Planter boxes and fishing nets
- Food service cups, cutlery, trays, and straws

## MAJOR POTENTIAL DISPOSAL ENVIRONMENTS

- Composting facilities or soil burial
- Anaerobic digestion
- Wastewater treatment facilities
- Plastics reprocessing facilities
- Landfill
- Marine and freshwater environments
- General open environment as litter

## **SOME ADVERSE ENVIRONMENTAL RISKS**

...

- Pollution in waterways due to high biochemical oxygen demand concentrations
- Migration of plastic degradation by-products to groundwater and surface water bodies
- Trauma and death of marine species
- Soil and crop degradation resulting from the use of compost that may have unacceptably high organic and or metal contaminants



**ASTM DEFINES  
BIODEGRADABLE  
AS**

capable of undergoing decomposition into carbon dioxide, methane, water, inorganic compounds, or biomass in which **the predominant mechanism is the enzymatic action of microorganisms**, that can be measured by standardized tests, in a specific period of time, reflecting available disposable conditions

## Experimental design

### POLYMERS

HDPE  
 PE-g-MA  
 PLA  
 PCL  
 Mater-Bi®

### COMPOUNDING

REACTIVE  
 EXTRUSION

### BLENDS

PLA60  
 PCL60  
 SPLA50  
 SPLA70  
 SPCL50



## Experimental design

Blend	HDP E	PE-g- MA	PLA	PCL	Mater-Bi®	Chemical formula
PLA 60	30	10	60	0	0	$C_3H_5O$
PCL 60	30	10	0	60	0	$C_5H_{10}O$
SPLA 50	30	10	0	0	60 (50 TPS + 50 PLA)	$C_5H_9O$
SPLA 70	30	10	0	0	60 (30 TPS + 70 PLA)	$C_3H_4O$
SPCL 70	30	10	0	0	60 (30 TPS + 70 PCL)	$C_3H_7O$

## Specific standard for biodegradable plastics

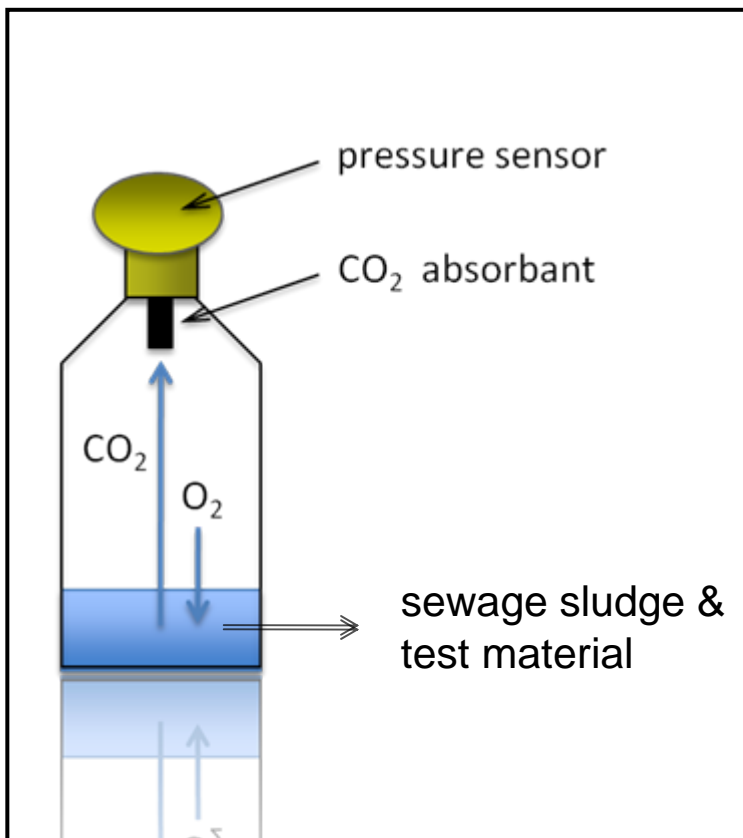
**ISO 14851 (1999)**

Determination of the ultimate aerobic biodegradability of plastic materials in an aqueous medium

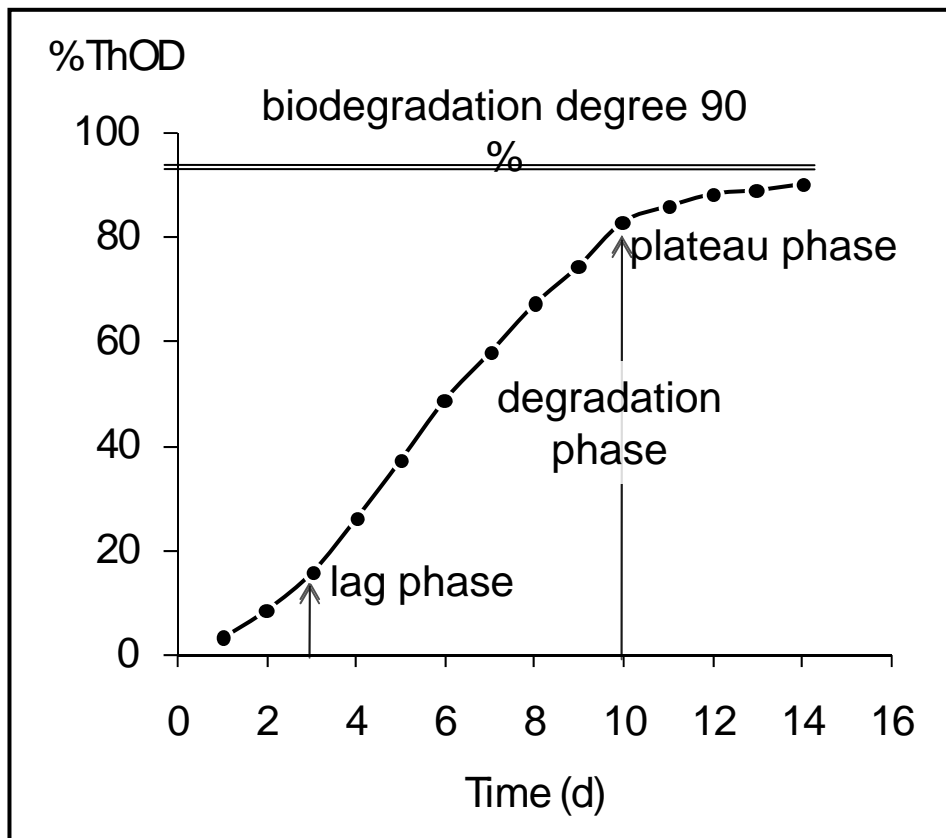
Conversion to carbon dioxide, water and biomass via microbial metabolism

# Oxygen consumption as a quantitative measure of biodegradation

## Oxitop®



## Biodegradation curve



## Saturation type mathematical model

### STATE VARIABLES

- X - biomass concentration
- BOD - biochemical oxygen concentration

### MODEL PARAMETERS

- $\mu_{\max}$  - maximum specific growth rate
- $k_s$  - affinity constant
- Y - biomass yield
- Z - conversion factor polymer to oxygen

### RATE EQUATIONS

Rate of microbial growth    Rate of polymer consumption    Rate of BOD formation

$$r_X = \frac{dX}{dt} = \mu_{\max} \frac{P}{k_s + P} X \quad r_P = \frac{dP}{dt} = -\frac{1}{Y} r_X = -\frac{\mu_{\max}}{Y} \frac{P}{k_s + P} X \quad r_{\text{BOD}} = \frac{1}{Z} r_P = \frac{\mu_{\max}}{YZ} \frac{\text{BOD}}{k_s + \text{BOD}} X$$

# First order mathematical model

## STATE VARIABLE

- BOD – biochemical oxygen concentration

## MODEL PARAMETERS

- $k_1$  – first order rate constant
- $BOD_u$  – ultimate BOD

## RATE EQUATION

$$\frac{dBOD_t}{dt} = k_1 (BOD_u - BOD_t)$$

# Mathematical model

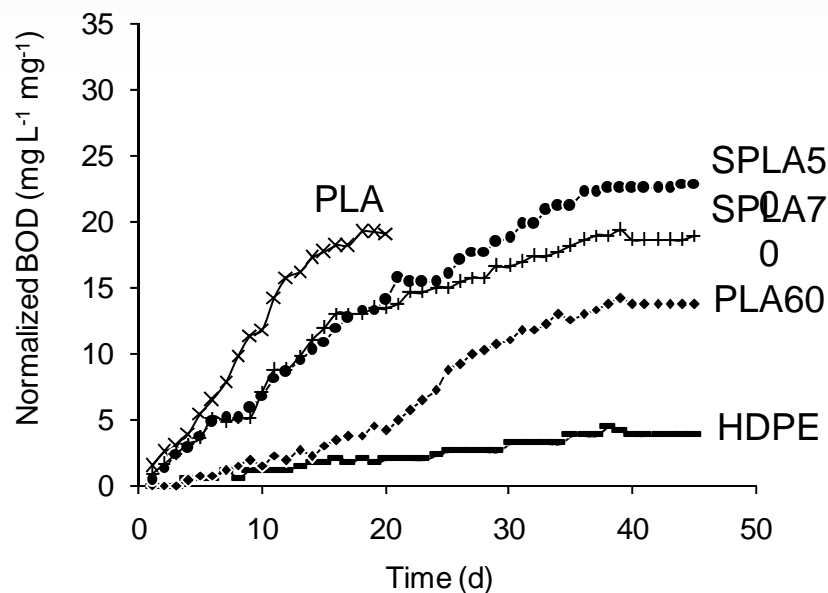
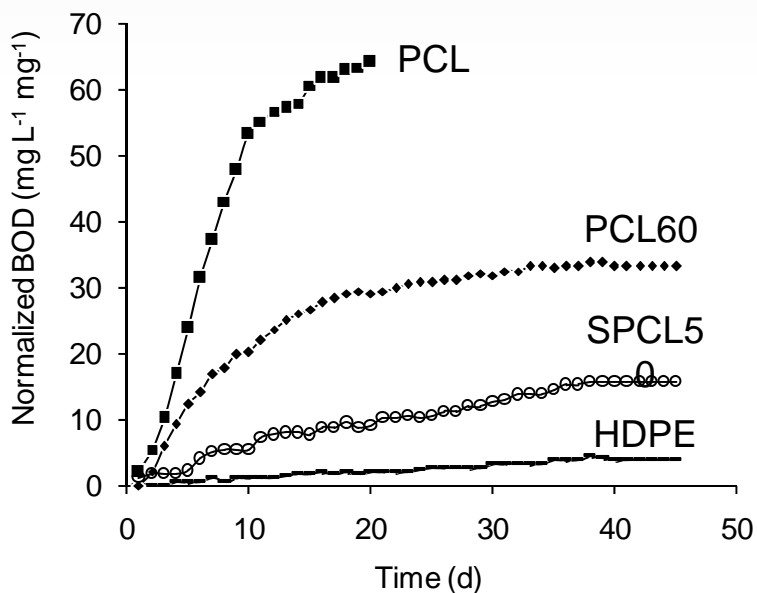
**METHOD**

non linear regression analysis

**SOFTWARE**

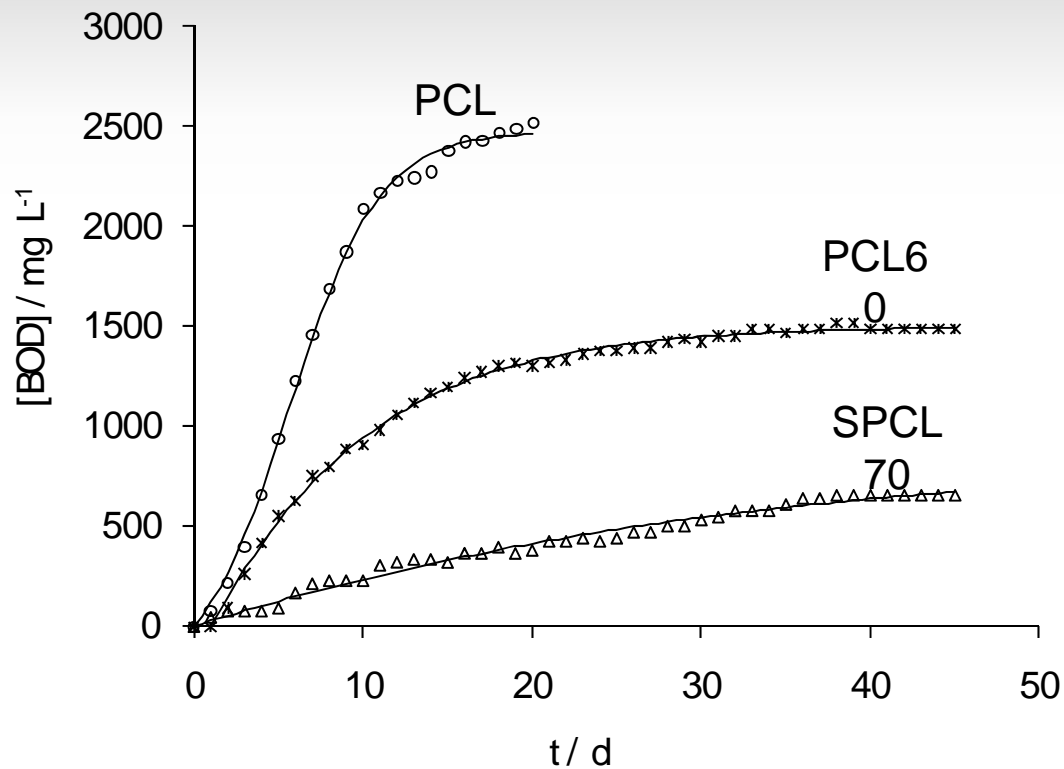
Berkeley Madonna

# Biodegradation kinetics



- PCL and PCL60 blend are more biodegradable than PLA and PLA60 blend
- Starch increased biodegradability of PLA blend; the same was not observed for the of PLC blend

**THE BIODEGRADABILITY TEST SUGGESTS THAT STARCH IS MORE BIODEGRADABLE THAN PLA BUT LESS THAN PCL**



### Sigmoidal curve

**PCL**

100 % ThOD  
 $\mu_{\max} = 0.43 \text{ d}^{-1}$

### Parabolic curves

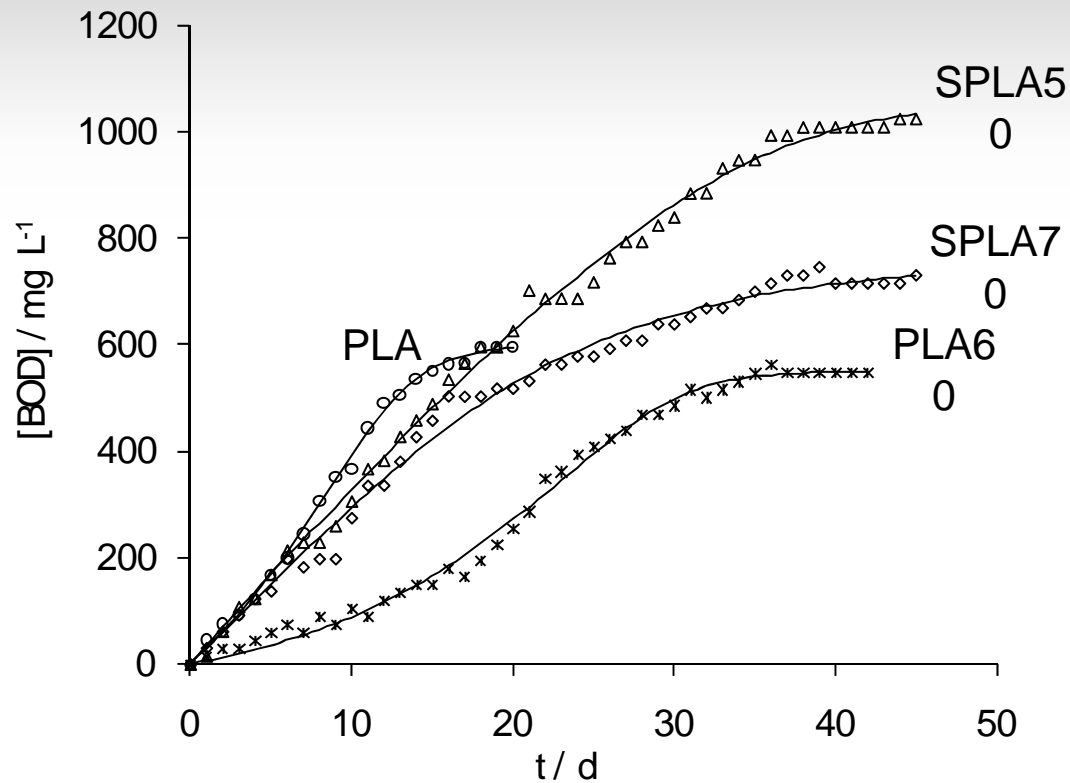
**PCL60**

48 % ThOD  
 $k_1 = 0.11 \text{ d}^{-1}$

**SPCL70**

24 % ThOD  
 $k_1 = 0.03 \text{ d}^{-1}$





### Sigmoidal curve

### Parabolic curves

**PLA**

**PLA60**

**SPLA50**

**SPLA70**

37 % ThOD

$\mu_{max} = 0.17 \text{ d}^{-1}$

22 % ThOD

$\mu_{max} = 0.08 \text{ d}^{-1}$

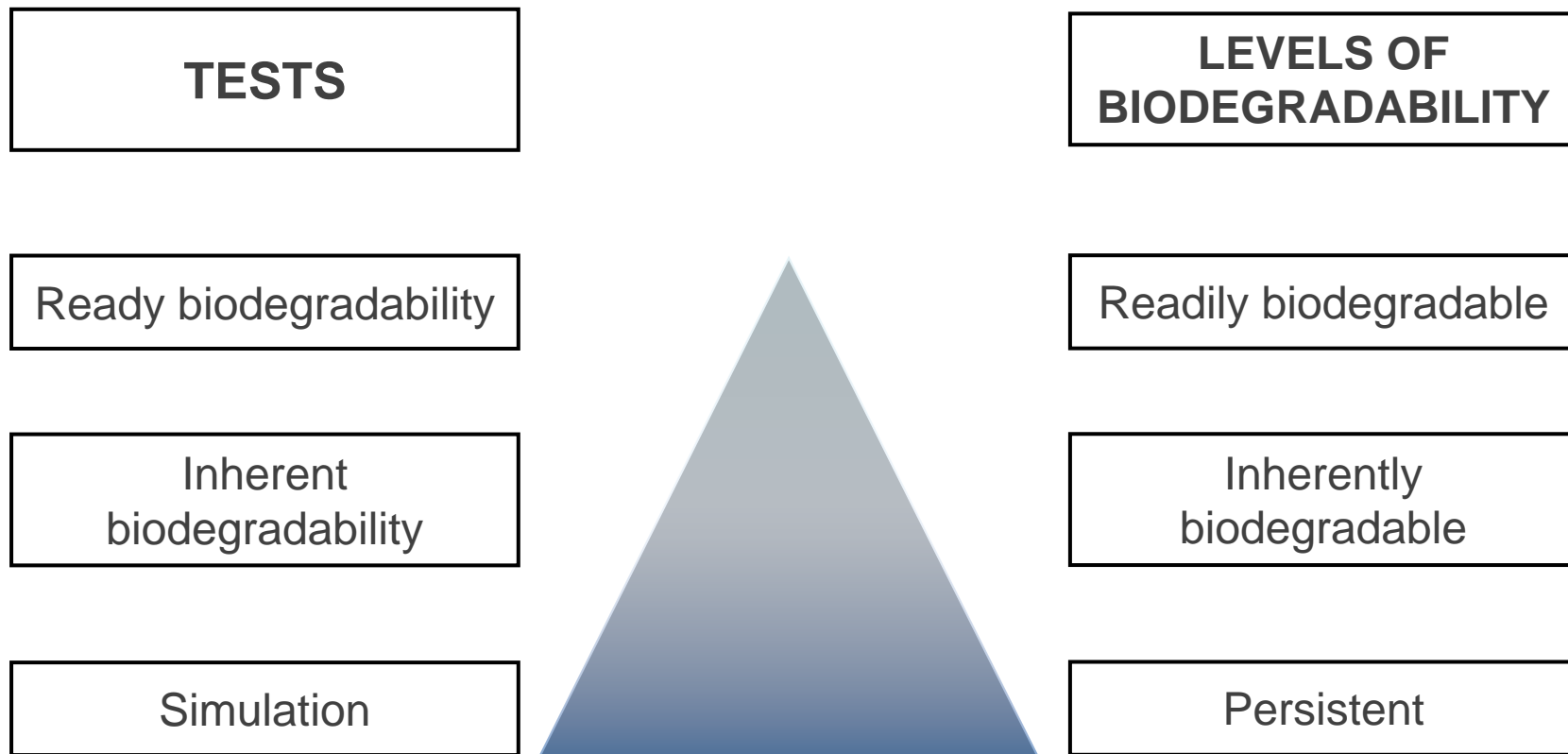
34 % ThOD

$k_1 = 0.02 \text{ d}^{-1}$

25 % ThOD

$k_1 = 0.03 \text{ d}^{-1}$

# Testing biodegradability (OECD)

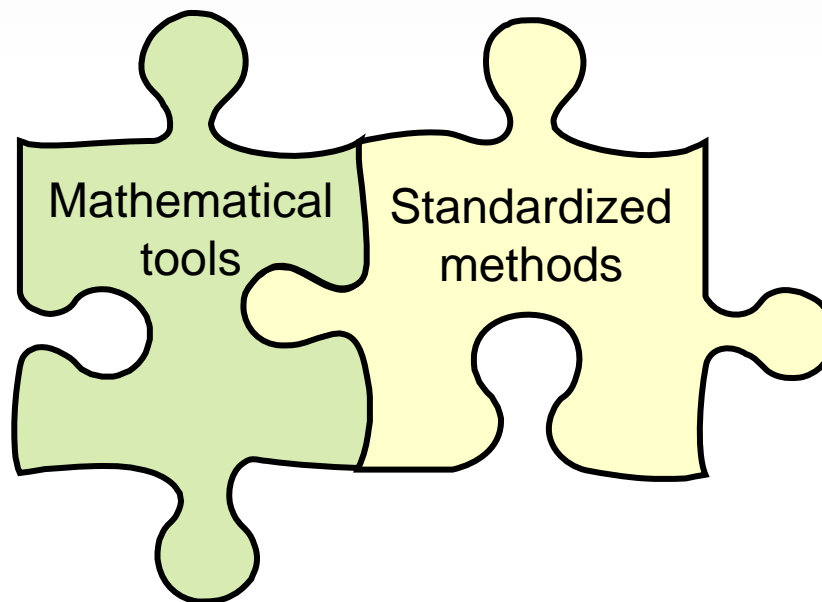


## Testing biodegradability (OECD)

**READILY  
 BIODEGRADABLE**

- lag phase - 10 % of the chemical is consumed
- degradation phase - extent of degradation exceeds 60 % for ThOD within 10 d window

**The blends did not pass the “ready”-level test because degradation occurs too slow**



- The first-order kinetic model could be used instead of the 10-day-window to describe degradation kinetics
- The saturation model could be appropriate in estimating the lag phase as well as the final extent of degradation